2005 Vol. 7, No. 6 1149–1151

## Fluoro, Alkylsulfanyl, and Alkylsulfonyl Leaving Groups in Suzuki Cross-Coupling Reactions of Purine 2'-Deoxynucleosides and Nucleosides<sup>†</sup>

Jiangqiong Liu and Morris J. Robins\*

Department of Chemistry and Biochemistry, Brigham Young University, Provo, Utah 84602-5700

morris\_robins@byu.edu

Received January 12, 2005

## **ABSTRACT**

Protected 2'-deoxynucleoside and nucleoside derivatives of 6-fluoropurine, 6-(3-methylbutyl)sulfanylpurine, and 6-(3-methylbutyl)ylsulfonylpurine undergo nickel- or palladium-mediated C-C cross-coupling with arylboronic acids to give good yields of 6-arylpurine products.

Modified purines and purine nucleoside derivatives play a major role in biochemistry and biology and as pharmaceutical agents.<sup>1</sup> Recently, 6-arylpurine ribonucleosides have been shown to exhibit cytostatic activity.<sup>2</sup> Classical methods for synthesis of nucleoside biaryls via Suzuki—Miyaura procedures have employed Pd- or Ni-mediated cross-couplings of aryl halides or sulfonates with arylboronic acids.<sup>3</sup> We recently reported a heteroaromatic Finkelstein process for conversion of 6-chloropurine 2'-deoxynucleoside and nucleoside deriva-

tives into the corresponding 6-iodopurine analogues and noted the markedly increased reactivity of the iodo compounds in certain classical organometallic cross-coupling reactions.<sup>4</sup> Aryl fluorides have rarely been used in such processes because of their diminished reactivity.

In 1981, we reported efficient methodology for the synthesis of 2-fluoropurine nucleosides by nonaqueous diazotization fluoro-dediazoniation of the 2-amino group of protected purine nucleosides.<sup>5</sup> Secrist et al. have also applied this protocol for the conversion of 6-amino- to 6-fluoropurine nucleoside derivatives.<sup>6</sup> We now report that this diazotative fluoro-deamination of protected adenosine and 2'-deoxy-adenosine analogues gives the 6-fluoropurine compounds in good yields. We then began an investigation of the utility of 6-fluoropurine nucleoside derivatives as cross-coupling

<sup>†</sup> Nucleic Acid Related Compounds. 125. Paper 124 is ref 9. (1) (a) Brathe, A.; Gunderesen, L.-L.; Rise, F.; Eriksen A. B.; Vollsnes, A. V.; Wang, L. *Tetrahedron* **1999**, *55*, 211–228. (b) Perez, O. D.; Chang Y.-T.; Rosania, G.; Sutherlin, D.; Schultz, P. G. *Chem. Biol.* **2002**, *9*, 475–

<sup>(2) (</sup>a) Hocek, M.; Holy, A.; Votruba, I.; Dvorakova, H. *J. Med. Chem.* **2000**, *43*, 1817–1825. (b) Hocek, M.; Holy, A.; Votruba, I.; Dvorakova, H. *Collect. Czech. Chem. Commun.* **2000**, *65*, 1683–1697. (c) Hocek, M.; Holy, A.; Votruba, I.; Dvorakova, H. *Collect. Czech. Chem. Commun.* **2001**, *66*, 483–499.

<sup>(3)</sup> Lakshman, M. K.; Thomson, P. F.; Nuqui, M. A.; Hilmer, J. H.; Sevova, N.; Boggess, B. *Org. Lett.* **2002**, *4*, 1479–1482.

<sup>(4)</sup> Liu, J.; Janeba, Z.; Robins, M. J. Org. Lett. 2004, 6, 2917-2919.

<sup>(5)</sup> Robins, M. J.; Uznanski, B. Can. J. Chem. 1981, 59, 2608–2611.
(6) Secrist, J. A.; Bennett, L. L.; Allan, P. W.; Rose, L. M.; Chang, C. H.; Montgomery, J. A. J. Med. Chem. 1986, 29, 2069–2074.

partners with arylboronic acids. We report that this opens an effective new avenue for modifications at C6 of purine nucleosides.

Our first challenge was to identify a catalytic complex that would insert readily into the purine C6–F bond. Several methods involving different transition metal centers have been described for activation of aromatic carbon–fluorine bonds. Cross-couplings of phenylmagnesium halides and fluorobenzenes have been performed at ambient temperature with nitrogen-heterocyclic carbene ligands and nickel catalysts.

We first tried Ni(COD)<sub>2</sub> with addition of 1,3-bis(2,6-diisopropylphenyl)imidazolin-2-ylidene (IPr) (Figure 1) for

Figure 1. Structure of the imidazolium-carbene ligand IPr.

attempted cross-coupling of 4-methoxyphenylboronic acid and 6-fluoro-9-[2,3,5-tri-O-(2,4,6-trimethylbenzoyl)- $\beta$ -D-ribo-furanosyl]purine. At ambient temperature, none of the coupling product was detected. However, we were delighted to find that the desired 6-(4-methoxyphenyl)-9-[2,3,5-tri-O-(2,4,6-trimethylbenzoyl)- $\beta$ -D-ribofuranosyl]purine (**1c**) was produced in high yield (84% isolated) in THF at 60 °C (Scheme 1) (Table 1). Different boronic acids were employed

**Scheme 1.** Couplings with 6-Fluoropurine Nucleosides

to evaluate the scope of this coupling reaction. Both electronrich and electron-poor arylboronic acids underwent coupling in good yields with 6-fluoropurine nucleoside derivatives. Application of this coupling protocol with a protected 6-fluoropurine 2'-deoxynucleoside also gave 6-arylpurine products in good isolated yields (Table 1).

Table 1. Yields of Coupling Products from Fluoropurines

entry	R	R'	Y	product (% yield)
1	Mes	OMes	Н	<b>1a</b> (84)
2	Mes	OMes	$\mathrm{CH}_3$	<b>1b</b> (82)
3	Mes	OMes	$OCH_3$	<b>1c</b> (84)
4	Mes	OMes	$\mathbf{F}$	<b>1d</b> (73)
5	Tol	H	$\mathrm{CH}_3$	<b>1e</b> (60)
6	Tol	H	$\mathbf{F}$	<b>1f</b> (67)

It is noteworthy that poor results were obtained upon replacement of Ni(COD)<sub>2</sub> by Pd(PPh<sub>3</sub>)<sub>4</sub> as the catalyst. With Pd(PPh<sub>3</sub>)<sub>4</sub>, major formation of an oxygen-insertion<sup>9</sup> compound **2** (Figure 2) was observed.

**Figure 2.** Structure of the oxygen-insertion compound 2.

We next focused our attention on cross-couplings of 6-alkylsulfanylpurine nucleoside derivatives, which are readily accessible by S<sub>N</sub>Ar displacements with 6-(imidazol-1-yl)-, <sup>10</sup> 6-(1,2,4-triazol-4-yl)-, <sup>11</sup> and 6-halopurine <sup>12</sup> precursors. They also are easily prepared by alkylation of thioinosine derivatives, <sup>12,13</sup> which can be obtained by deoxygenative thiation of inosine or deaminative sulfhydrolysis of 6-N-substituted adenosine intermediates. <sup>12</sup> Cross-coupling of Grignard reagents and 6-(methylsulfanyl)purine derivatives with a nickel—phosphine complex had been reported. <sup>14</sup>

Our first cross-coupling of 6-[(3-methylbutyl)sulfanyl]-9-(2,3,5-tri-O-acetyl- $\beta$ -D-ribofuranosyl)purine and 4-methoxyphenylboronic acid was incomplete after 8 h with Pd(OAc)<sub>2</sub>/IPr/K<sub>2</sub>CO<sub>3</sub>/THF at 60 °C. However, when the solvent was changed from THF to toluene and the temperature was

Scheme 2. Couplings with Sulfanylpurine Nucleosides

1150 Org. Lett., Vol. 7, No. 6, 2005

<sup>(7) (</sup>a) Braun, T.; Foxon, S. P.; Perutz, R. N.; Walton, P. H. *Angew. Chem., Int. Ed.* **1999**, *38*, 3326–3329. (b) Mongin, F.; Mojovic, L.; Guillamet, B.; Trecourt, F.; Queguiner, G. *J. Org. Chem.* **2002**, *67*, 8991–8994. (c) Kim, Y. M.; Yu, S. *J. Am. Chem. Soc.* **2003**, *125*, 1696–1697. (d) Widdowson, D. A.; Wilhelm, R. *Chem. Commun.* **2003**, 578–579. (8) Bohm, V. P. W.; Gstottmayr, C. W. K.; Weskamp, T.; Herrmann, W. A. *Angew. Chem., Int. Ed.* **2001**, *40*, 3387–3389.

increased to 90  $^{\circ}$ C, the coupling reaction was complete in 8 h. Electron-rich and electron-poor arylboronic acids were also well tolerated with the alkylsulfanylpurine substrates (Scheme 2; Table 2).

Table 2. Yields of Coupling Products from Sulfanylpurines

entry	R	Y	product (% yield)
1	Tol	$\mathrm{CH}_3$	<b>3a</b> (69)
2	Ac	$OCH_3$	<b>3b</b> (78)
3	Tol	$\mathbf{F}$	<b>3c</b> (71)

The oxidation state of the sulfur substituent at C6 was then briefly probed. Oxidation of 6-benzylsulfanylpurine nucleoside derivatives with Oxone in buffered brine had given 6-benzylsulfonyl compounds in high yields.  $^{10}$  Oxidation of a protected 6-(isopentylsulfanyl)purine nucleoside gave 6-[(3-methylbutyl)sulfonyl]-9-[2,3,5-tri-O-(2,4,6-trimethylbenzoyl)- $\beta$ -D-ribofuranosyl]purine (4). The sulfone 4 and 4-methoxyphenylboronic acid underwent coupling at 60 °C in 8 h [Pd(OAc)\_2/IPr/K\_3PO\_4/THF] to give 1c (81%, isolated yield) (Scheme 3).

This coupling with the sulfone 4 occurred more readily (60 °C, THF) than with the corresponding thioether (90 °C, toluene). It was known that arylsulfonyl chlorides function

**Scheme 3.** Coupling with a Sulfonylpurine Nucleoside

as substrates for Suzuki and Stille couplings, 15 but we did not find prior examples of Suzuki couplings with sulfones.

In summary, we have developed nickel- and palladium-based systems with imidazolium-carbene ligands that catalyze efficient Suzuki cross-couplings of arylboronic acids and 6-fluoro-, 6-[(3-methylbutyl)sulfanyl]-, and 6-[(3-methylbutyl)sulfonyl]purine nucleoside derivatives to give the corresponding 6-arylpurine products. These reactions enlarge the scope of our complementary Suzuki couplings of arylboronic acids and 6-(azolyl)purine derivatives and expand possibilities for new medicinal applications.

**Acknowledgment.** We gratefully acknowledge pharmaceutical company gift funds in support of this research (M.J.R.) and the award of a Roland K. Robins Graduate Research Fellowship (J.L.) by Brigham Young University.

**Supporting Information Available:** Experimental procedures, characterization data, and <sup>1</sup>H and <sup>13</sup>C NMR spectra for new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

## OL050063S

(15) (a) Dubbaka, S. R.; Vogel, P. J. Am. Chem. Soc. **2003**, 125, 15292–15293. (b) Dubbaka, S. R.; Vogel, P. Org. Lett. **2004**, 6, 95–98.

Org. Lett., Vol. 7, No. 6, 2005

<sup>(9)</sup> Liu, J.; Robins, M. J. Org. Lett. 2004, 6, 3421-3423.

<sup>(10)</sup> Lin, X.; Robins, M. J. Org. Lett. 2000, 2, 3497-3499.

<sup>(11) (</sup>a) Samano, V.; Miles, R. W.; Robins, M. J. J. Am. Chem. Soc. **1994**, 116, 9331–9332. (b) Miles, R. W.; Samano, V.; Robins, M. J. J. Am. Chem. Soc. **1995**, 117, 5951–5957.

<sup>(12)</sup> Srivastava, P. C.; Robins, R. K.; Meyer, R. B., Jr. In *Chemistry of Nucleosides and Nucleotides*; Townsend, L. B., Ed.; Plenum: New York, 1988; Vol. 1, pp 113–281.

<sup>(13) (</sup>a) Lin, T.-S.; Cheng, J.-C.; Ishiguro, K.; Sartorelli, A. C. *J. Med. Chem.* **1985**, 28, 1481–1485. (b) Gupte, A.; Buolamwini, J. K. *Bioorg. Med. Chem. Lett.* **2004**, *14*, 2257–2260.

<sup>(14)</sup> Sugimura, H.; Takei, H. Bull. Chem. Soc. Jpn. 1985, 58, 664-666.